

Institute for Spectrometry and Radiation Protection

Logistic, Consulting & Management for $\alpha\cdot\beta\cdot\gamma\cdot n$ –Spectrometry and Radiation Protection. Reference Systems and Software Products for *In Vivo, In Vitro* and *In Situ* Application. Service for α -, β -, γ -, & n-Spectrometry Systems

Laboratory - Radionuclide Handbook

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Now with a separate large energy (~2800 photon-lines) and isotope list (~290 Radioisotopes)

Content – Quick-Info, Vorwort/Preface, Introduction, Notes, Descriptions, Symbols

Anthropogenic

- > Activation & fission products
- NucMed: therapy, diagnostic, PET, ...
 Dose: emitted energies per decay
- ➤ Iodine's & special nuclides
- Nuclide boxes with all decay details

Actinides

- \triangleright U/Th (n,2n) & (n, γ) activation process
- Pu/Am/Np/U/Th/Pa production
- ➤ All four Plutonium decay series
- Nuclide boxes with all decay details

Naturals

- ➤ All decay series
- > U-238/Th-232/U-235/Np-237/Pu-244
- > Isotopics: Uranium ratios
- Cosmogenic, complementary nuclides
- Nuclide boxes with all decay details

Particle Induced Radiation

- Shielding, detector/mounting material
- Muon induced radiation
- Neutron scattering & capture
- Activation isotopes
- Fluorescence in shielding materials

Calibration Radioisotopes

- > Standard & multiphoton sources
- ➤ Low & high energy isotopes
- Natural intrinsic check sources
- Neutron sources as Cf & Am

Spectrometry Source-Detection Effects

- SE/DE/XE: single/double/x-ray escape
- CS/RS: coincidence/random summing
- Line broadening (Li-7)
- > Asymmetric peaks: Ge lattice recoil
- ➤ Air filter samples: Pb-212 x-ray triplets

| reference | β -, ε -decay, IT | β -, ε -, IT- | daughter-nuclide & E_{γ} | | γ- & X- |
|--|---|---------------------------------|--|-----------------------------|--|
| isotope | $E_{\beta}^{max}, (\bar{E}_{\beta})$ | branch | half-life $(t_{1/2}^m)$ for IT levels | E-list in | intensity absol. |
| $Q_{\varepsilon}, Q_{\beta} \text{ MeV}$ | , | intensity | γ -ray transitions to final state | I-order | equil./absolute |
| $t_{1/2}$ | MeV | % | in keV, dose in MeV/Bq-s | keV | % |
| NDS:108,2173,2007 | | $\beta^{-}:100\%$ | γ -/ β ⁻ -dose: 0.5631 ₁₃ /0.1871 ₁₂ | in equilibrium | with ¹³⁷ Ba |
| ¹³⁷ ₅₅ Cs | $\rightarrow \beta^{-}$: 1.17563 ₁₇ | 5.30_{20} | $\rightarrow {}^{137}_{56} \mathrm{Ba}$ (abundance: 11.232%) | | |
| | $\rightarrow \beta^{-}$: 0.89213 ₂₀ | 5.8 ₃ E-4 | $\rightarrow \gamma$: 283.5 | 283.5_{1} | 5.8 ₈ E-4 |
| | $\rightarrow \beta^{-}$: 0.51397 ₁₇ | 94.70_{20} | $\rightarrow ^{137m}_{56}$ Ba $\stackrel{IT}{\underset{2,55}{\longrightarrow}} ^{137}_{56}$ Ba | 661.657 ₃ | 85.10 ₂₀ (equilibr.) |
| | (0.1871_{10}) | | | | |
| $^{137m}_{56}{ m Ba}$ | \rightarrow IT: | 100 | $\rightarrow \gamma: 661.66^{M4,int} \rightarrow ^{137}_{56} \text{Ba}$ | 661.657 ₃ | 89.90 ₁₄ (absolute) |
| 2.552 ₁ m | | | $\rightarrow X: \Sigma K_{\alpha}$ | 32.06 | 5.63/5.94 |
| | | | $\rightarrow X: K_{\alpha_1}$ | 32.194 | 3.64 ₁₀ /3.84 ₁₀ |
| | Note: data set | | $\rightarrow X: K_{\alpha_2}$ | 31.817 | 1.995/2.106 |
| | completely | | $\rightarrow X: \Sigma K_{\beta}$ | 36.40 | 1.2335/1.310 |
| | | | $\rightarrow X: K_{eta_1}$ | 36.358 | $0.672_{18}/0.709_{18}$ |
| | | | $\rightarrow X: K_{\beta_3}$ | 36.304 | 0.3489/0.36710 |
| $Q_{\beta^-}=1.17563_{17}$ | | | $\rightarrow X: K_{eta_2}$ | 37.255 | 0.2136/0.2446 |
| 30.08 ₉ y | $\alpha_t^{661} = 0.1102_{19}$ | | $\rightarrow X: \Sigma L$ | 4.47 | 0.9134/0.975 |
| $int = {}^{60}Co(662.24,$ | SE); ²⁴¹ Am(662.40, 0 | x-ray intensity | ‡equil./absol.‡ | | |

Additional: $\alpha/\beta/\gamma$ -energies in E-order for nuclide each, U-isotopics, coincidence corrections, comments to data evaluation, list of isotopes & materials, references, nuclide & keyword index.

Example for Isotope Production and Decay of Actinides

Production of Pu-isotopes via U- and Pu-239 in a nuclear reactor:

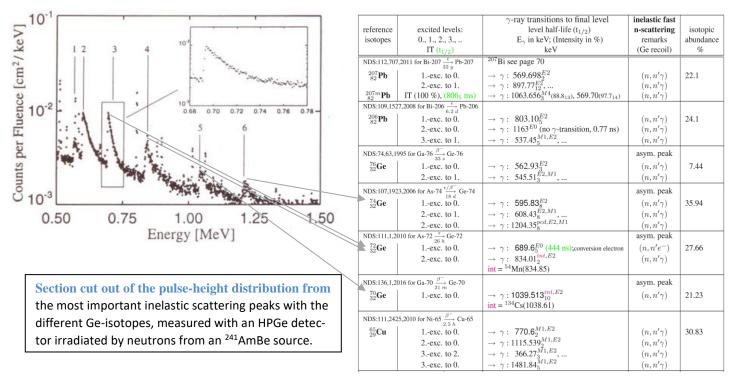
 $\frac{238}{92}$ U + $\frac{1}{0}$ n $\longrightarrow \frac{239}{92}$ U $\frac{\beta^{-}}{23.5m}$ $\frac{239}{93}$ Np $\frac{\beta^{-}}{2.36d}$ $\frac{239}{94}$ Pu(24,110y) $\frac{(n,\gamma)}{269b}$ Pu-240 up to Pu-246

Decay of Am-243/Pu-239 down to natural U-235

| $ \begin{array}{ccc} 47 \mathbf{Pu} & \xrightarrow{\beta^{-}} \\ 100\% & & \\ 2.27 & d & & \\ \end{array} $ | ²⁴⁷ ₉₅ Am 23.0 m | $\xrightarrow{\beta^-}_{100\%}$ | $^{247}_{96}$ Cm 1.56*10 ⁷ y 100 % $\psi \alpha$ $^{243}_{94}$ Pu | $\xrightarrow{\beta^-}$ | $^{247}_{97}$ Bk $^{1380}_{97}$ y $^{100}_{95}$ $^{4}_{05}$ $^{243}_{95}$ Am | € 99.965% | $^{247}_{98}$ Cf 3.11 h 0.035 % $\psi \alpha$ $^{243}_{96}$ Cm | | | half life t _{1/2} 4.956 h | isotope decay branch, % $\frac{243}{94} Pu^{**}$ $\frac{100}{\beta} \searrow$ | α-decay energy, MeV (intensity, %)** | β-decay energy, MeV (intensity, %)** β: 0.579 (60) β: 0.495 (21.0) β: 0.537 (8.9) | γ-emission energy, keV; (intensity, %)* γ: 84.0 (23.1) γ: 25.2 (8.0) γ: 41.8 (0.76) |
|---|--|--|--|--|--|---------------------------------|--|--|--|--|---|--|---|--|
| | | | 4.956 h | 100% | 7370 y 100 % ↓ α | 0.29% | 29.1 y 99.71 % ↓ α | | | 23.45 m 7370 y | $^{239}_{92}$ U** $^{243}_{95}$ Am $^{100}_{100}$ $^{100}_{\beta}$ $^{\downarrow}$ $^{\alpha}$ | α: 5.275 (87.1) _{Am} $α$: 5.233 (11.2) _{Am} | β : 1.189 (69.0) _U β : 1.264 (18.7) _U β : 1.232 (9.4) _U | γ : 74.66 (49.2) _U γ : 43.53 (4.07) _U γ : 74.66 (67.2) _{Am} γ : 43.53 (5.90) _{Am} |
| Isotope produ (n,γ) -reaction and ^{239}Pu : fol | via ^{239}U | U $\xrightarrow{-238}$ n,γ σ =2.7b | ²³⁹ ₉₂ U 23.45 m | $\xrightarrow{\beta^-}$ 100% | ²³⁹ ₉₃ Np 2.3565 d | $\xrightarrow{\beta^-}$ 100% | 239 Pu 24110 y | $\stackrel{n,\gamma}{\underset{\sigma=269b}{\longrightarrow}}$ | $ \begin{array}{c c} Pu-240 \\ - \\ Pu-246 \end{array} $ | 7.2 m 2.3565 d | $^{235}_{90}$ Th** $^{239}_{93}$ Np $^{100}_{}$ $^{100}_{}$ $^{}\beta\downarrow$ $^{}\downarrow\beta$ | Th: no more data available | β: 0.436 (45) β: 0.336 (42.0) β: 0.392 (7.0) | γ : 106.12 (26.3) _{Np} γ : 277.60 (14.44) _{Np} γ : 228.18 (11.14) _{Np} Th- γ : no data available |
| 4. 10 | tow cyun | 0-2.70 | | 3- | | a- | 100 % ↓ α | | | 24.44 m 24110 y | $^{235}_{91}$ Pa $^{239}_{94}$ Pu** 100 100 $^{\beta} \searrow ^{\prime} ^{\alpha}$ | α: 5.157 (70.77) α: 5.144 (17.11) α: 5.106 (11.94) | β: 1.41 (100) β: 1.03 (0.005) β: 0.72 (0.005) | γ : 51.62 (0.0272) $_{Pu}$ γ : 38.66 (0.0104) $_{Pu}$ Pa- γ : no reliable |
| | | | ²³⁵ ₉₀ Th 7.2 m | $\stackrel{\beta^-}{\longrightarrow}$ 100% | ²³⁵ ₉₁ Pa 24.5 m | $\xrightarrow{\beta^-}$ 100% | 235 ₉₂ U 7.038*10 ⁸ y 100 % | € 99.9974% | ²³⁵ ₉₃ Np 396.1 d 0.0026 % | 7.038*10 ⁸ y | ²³⁵ U** ⁹² 100 ↓ α | a: 4.395 (57.73) a: 4.364 (18.92) | | data available γ: 185.72 (57.0) γ: 143.76 (10.96) γ: 163.36 (5.08) |
| | | | pe production | | $0.13b, 7MeV \\ 2.1b, 10MeV$ | Th-232 $n,2n$ | $\psi \alpha$ $^{231}_{90}$ Th | $\frac{\beta^-}{100\%}$ | $\psi \alpha$ $\begin{array}{c} 231 \text{Pa} \\ 91 \end{array}$ | 25.52 h | $^{231}_{90}$ Th** 100 | | β: 0.290 (40) β: 0.307 (32) β: 0.208 (12.1) | γ: 25.64 (14.1) γ: 84.21 (6.6) γ: 89.95 (1.00) |
| | | | follow <mark>brown</mark> | | | | 25.52 h | n,γ ✓ | 3.276*10 ⁴ y 100 % | 3.276*10 ⁴ y | $^{231}_{91}$ Pa 100 $\Downarrow \alpha$ | α: 5.014 (25.4) α: 4.951 (22.8) α: 5.028 (20) | β: 0.289 (12) | γ: 27.36 (10.3) γ: 302.65 (2.88) γ: 300.07 (2.47) |
| | | | | | Pa-231 c | ontinue pa | Pa-234m age 12 and de | σ=210 <i>b</i> etails 23 | $\Downarrow \alpha$ | | emission: intensity per l double asterix are act | | | $\Sigma \approx 100\%$ (d,2n) interacting processes |

Example for Partical Induced Radiation

Neutron scattering with asymmetric photon peak-shape: the n-spectrum and the isotope list are showing the Ge-isotopes



Detector response: The contribution to the full energy deposition of the photon or conversion electron emitted in the reaction leads to narrow gauss peaks whereas the fraction of kinetic energy transferred to the recoil nucleus or to secondary elastic scattering leads to exponential decreasing peak responses caused by superposed spectra of mono-energetic neutrons.

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